

Name: _____ Period: _____

Practicing Drawing Atoms...

For each element please determine the number of protons, neutrons, and electrons, and then draw the atom. Use a (+) sign to represent the protons, (0) sign to represent the neutrons, and a (-) sign to represent the electrons.

Example:

16
S
Sulfur
32.06

Protons = _____

Electrons = _____

Neutrons = _____

Draw the Atom:

1.)

4
Be
Beryllium
9.01

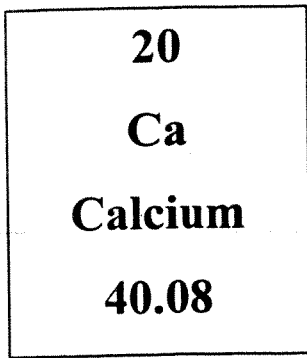
Protons = _____

Electrons = _____

Neutrons = _____

Draw the Atom:

2.)



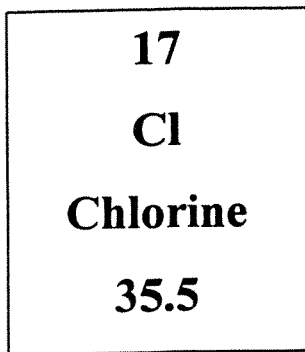
Protons = _____

Electrons = _____

Neutrons = _____

Draw the Atom:

3.)



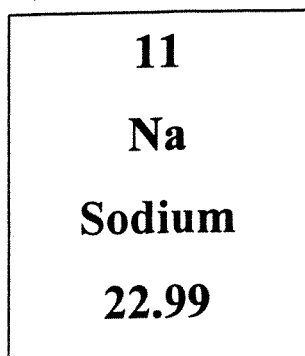
Protons = _____

Electrons = _____

Neutrons = _____

Draw the Atom:

4.)



Protons = _____

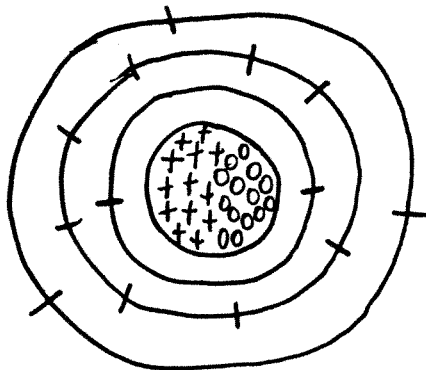
Electrons = _____

Neutrons = _____

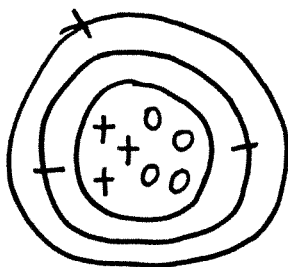
Draw the Atom:

Please look at the following atoms I have draw for you...please tell me what ELEMENT NAME and SYMBOL it represents.

5.) _____



6.) _____



7.) _____

